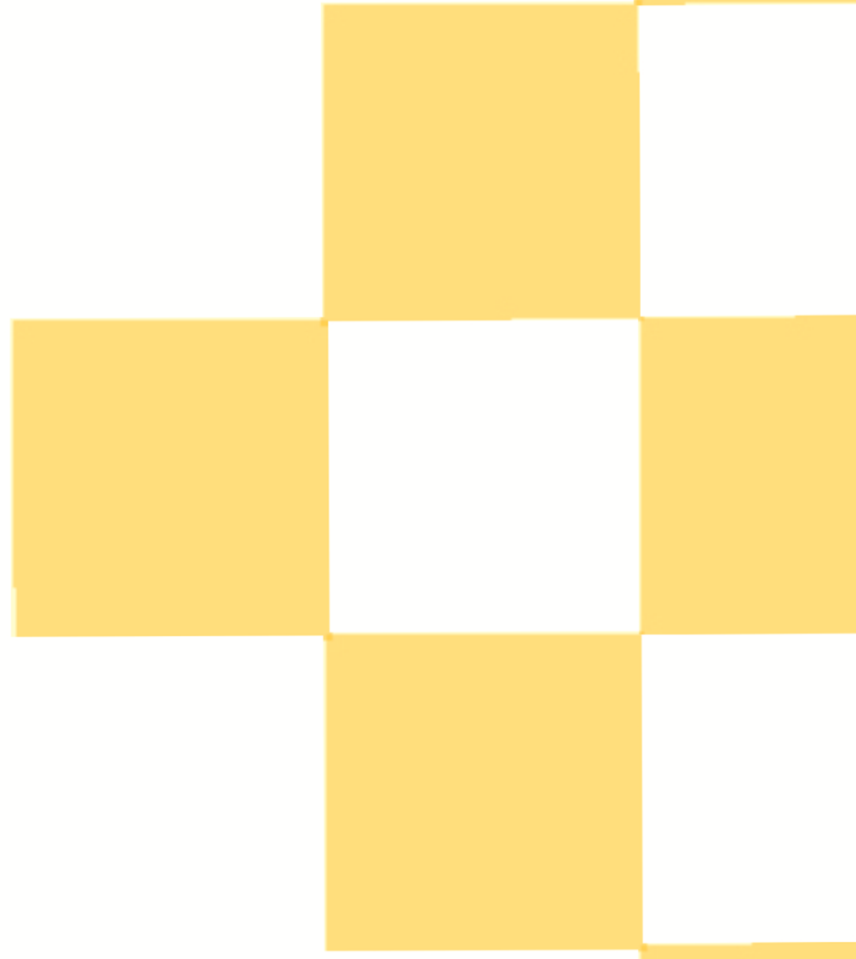


Structural and Raman spectroscopic characterization of C-TiO₂ nanotubes synthesized by a template-assisted sol-gel technique.

^{1,2}Nwabisa Takata , ¹Edson L. Meyer , ¹Raymond Taziwa, ¹David Katwire and ²Clive Oliphant

¹Institute of Technology, University of Fort Hare, Private Bag x1314, Alice 5700, South Africa

²National Metrology Institute of South Africa, CSIR Campus, Building 5, Meiring Naude Road, Brummeria, Pretoria 0182 South Africa.



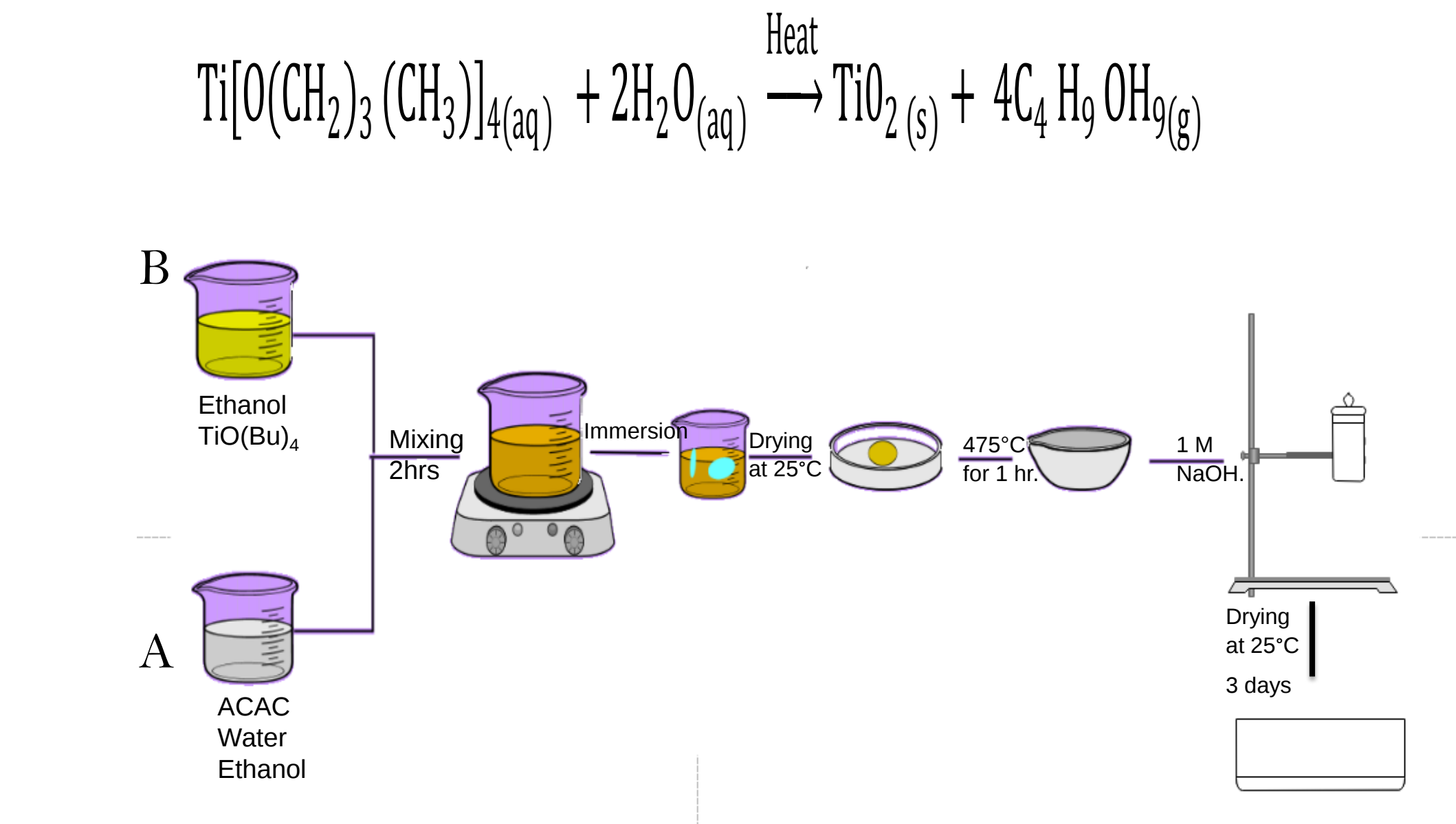
INTRODUCTION

- Continuous breakthroughs in synthesis of TiO₂ nanomaterials have brought applications such as photo-catalysis, photovoltaics (PV), etc.
- In PV TiO₂ has been used as a semiconductor material for dye sensitized solar cells (DSSC).
- DSSC photo-electrode utilizes TiO₂ nanoparticles, resulting to low efficient collection of photo-generated electrons due to electron trapping sites and lack of response in the UV vis region.
- Carbon doped nanotubes are a promising solution to spectral losses .

OBJECTIVES

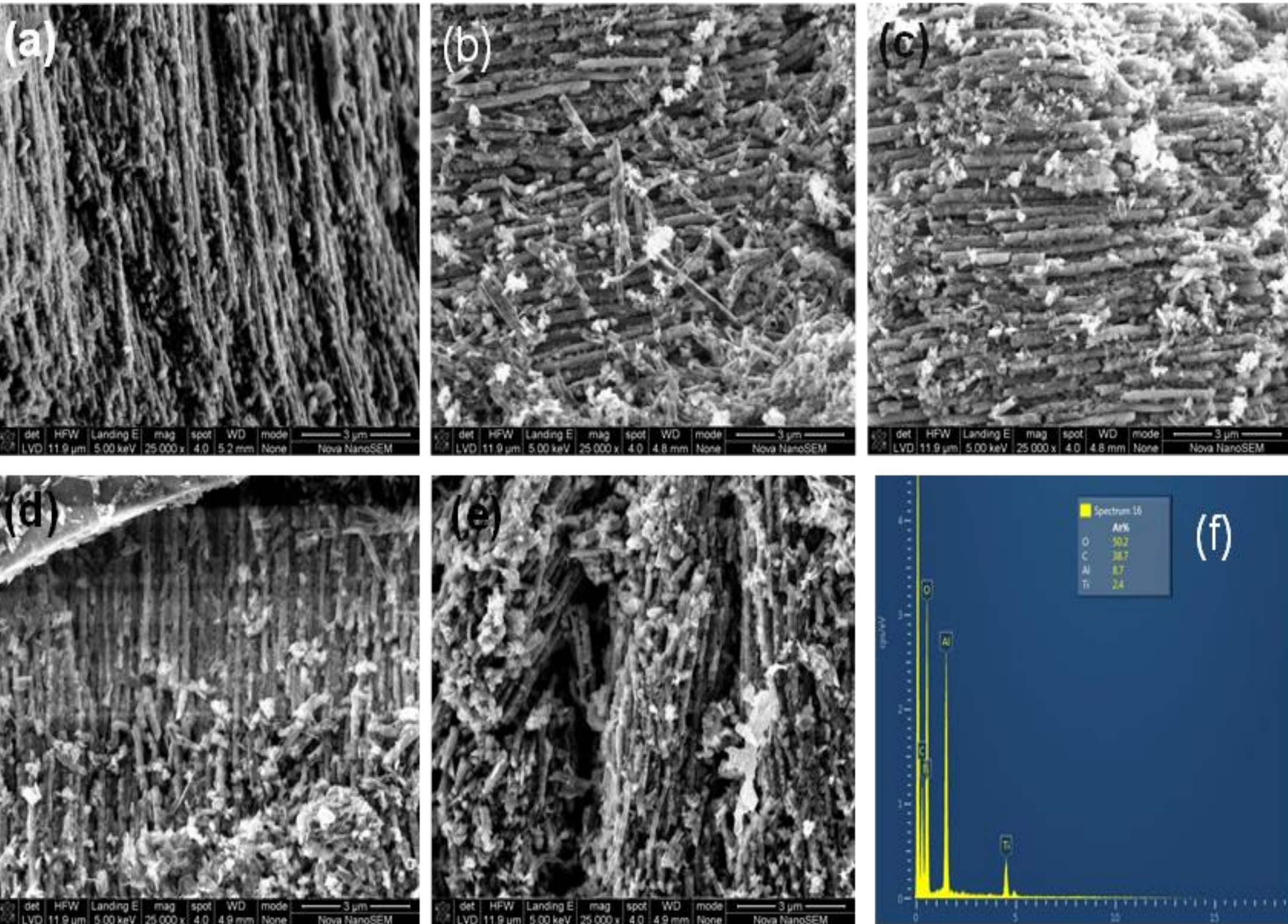
- Synthesis of undoped and carbon doped TiO₂ sol-gel precursor solutions.
- Synthesis of undoped and carbon doped TiO₂ nanotubes using AAM templates.
- Structural/ morphological, elemental and optical characterization of undoped and carbon doped TiO₂ nanotubes using SEM, SEM-EDX, FTIR, XRD, and Confocal Raman spectroscopy

Template-assisted sol-gel technique



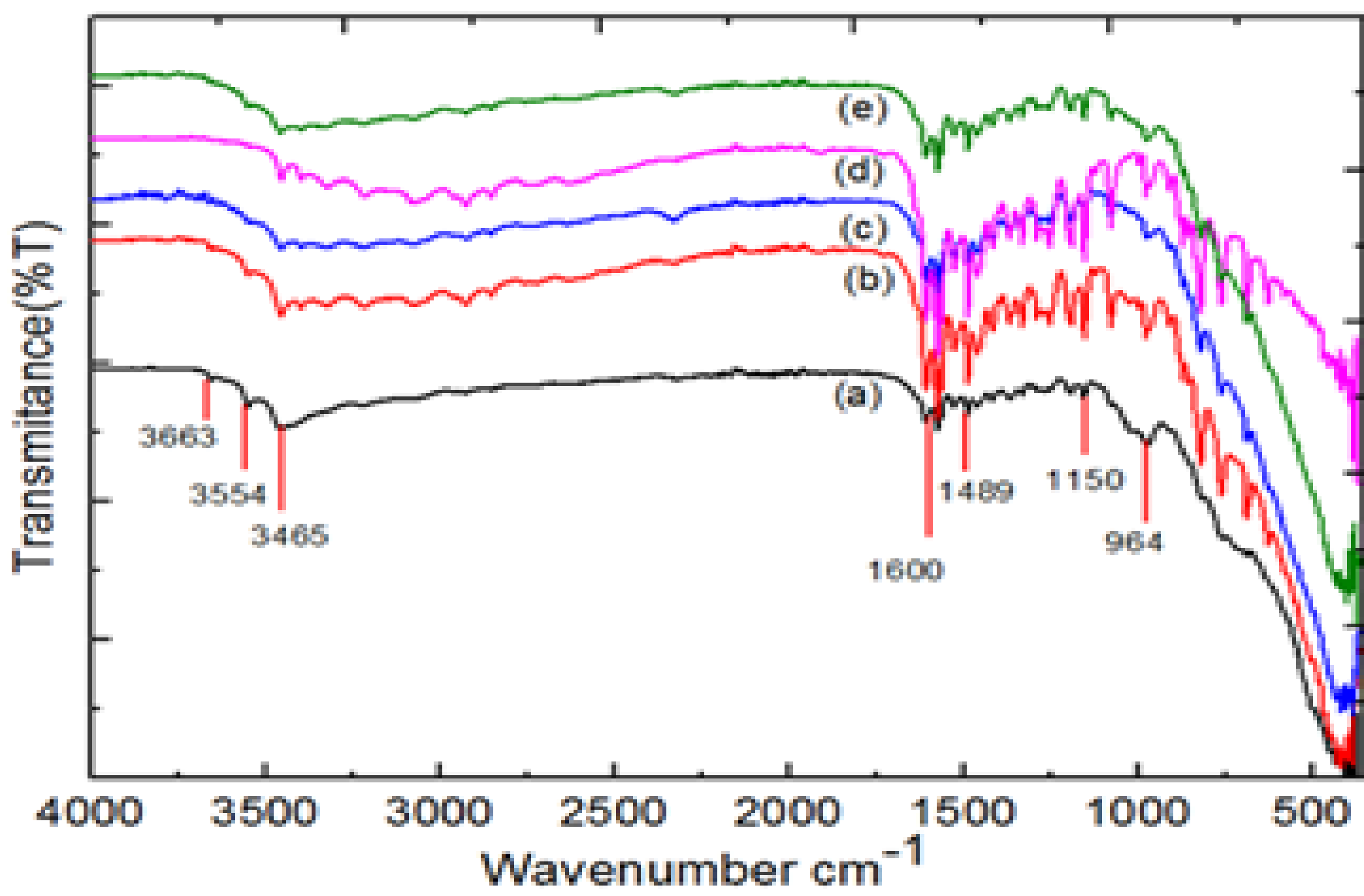
RESULTS AND DISCUSSION

Scanning Electron Microscopy



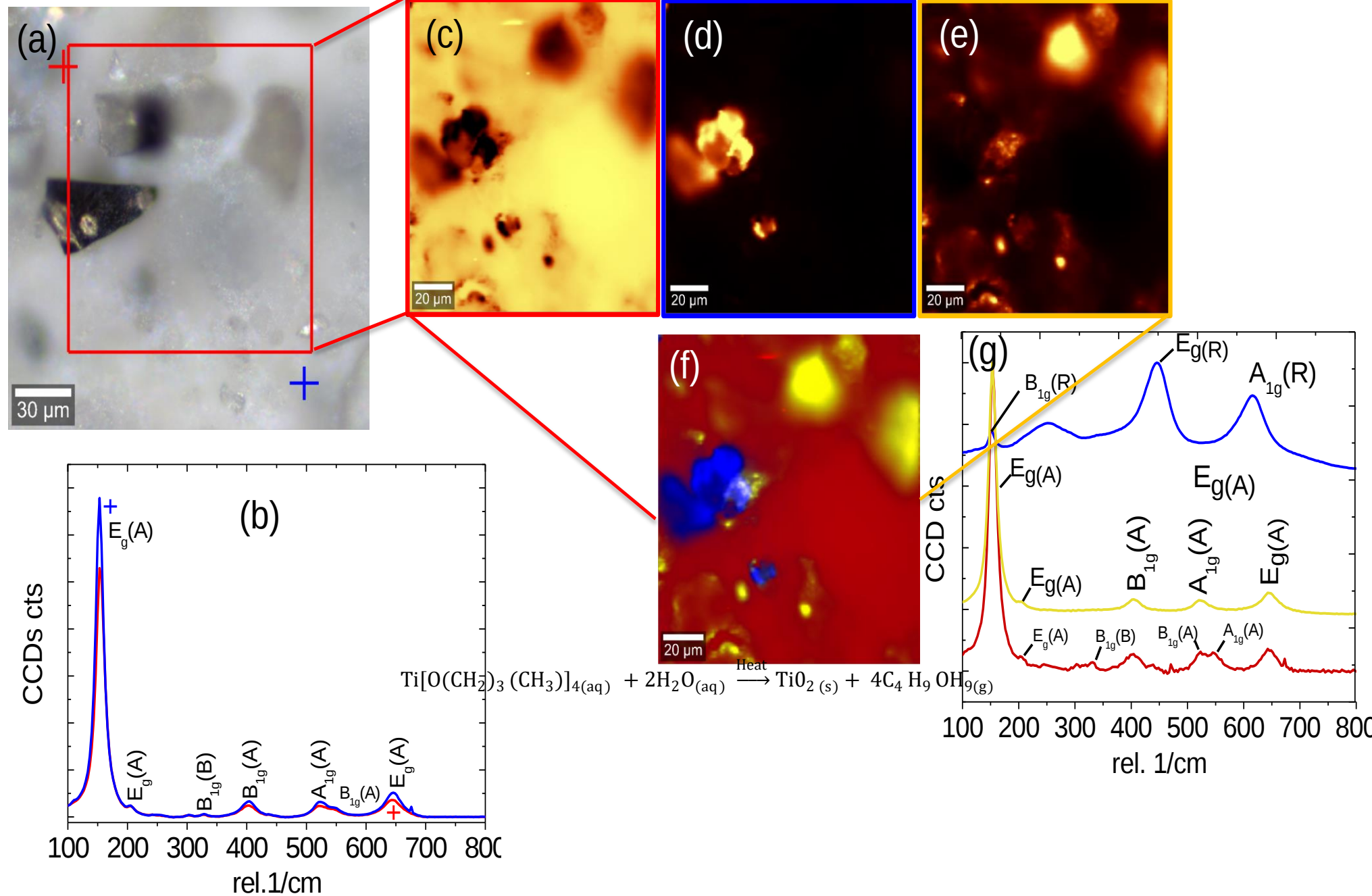
- SEM revealed that the un-doped TNTs are closely-packed in the form of bundles.
- Upon doping the TNTs become loosely packed
- EDX shows presence of Ti and O in the spectra confirms the successful hydrolysis of (Ti (OBU) 4)

Flourier Transform Infrared Spectroscopy



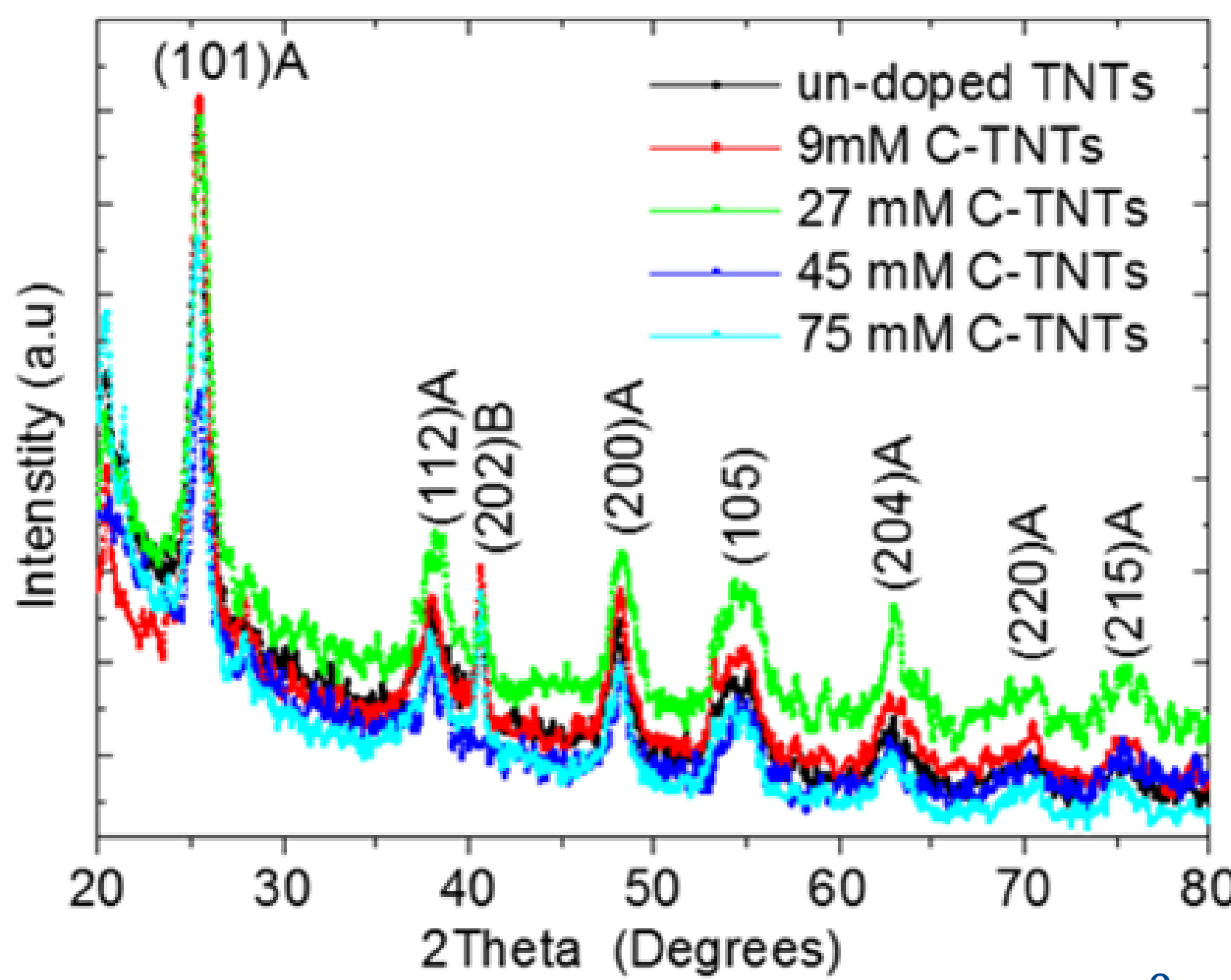
- FTIR Reveals a characteristic vibrational mode in the region 580-660 cm⁻¹ which is associated with Ti-O bonds in TiO₂ for all the synthesized samples

Raman Spectroscopy



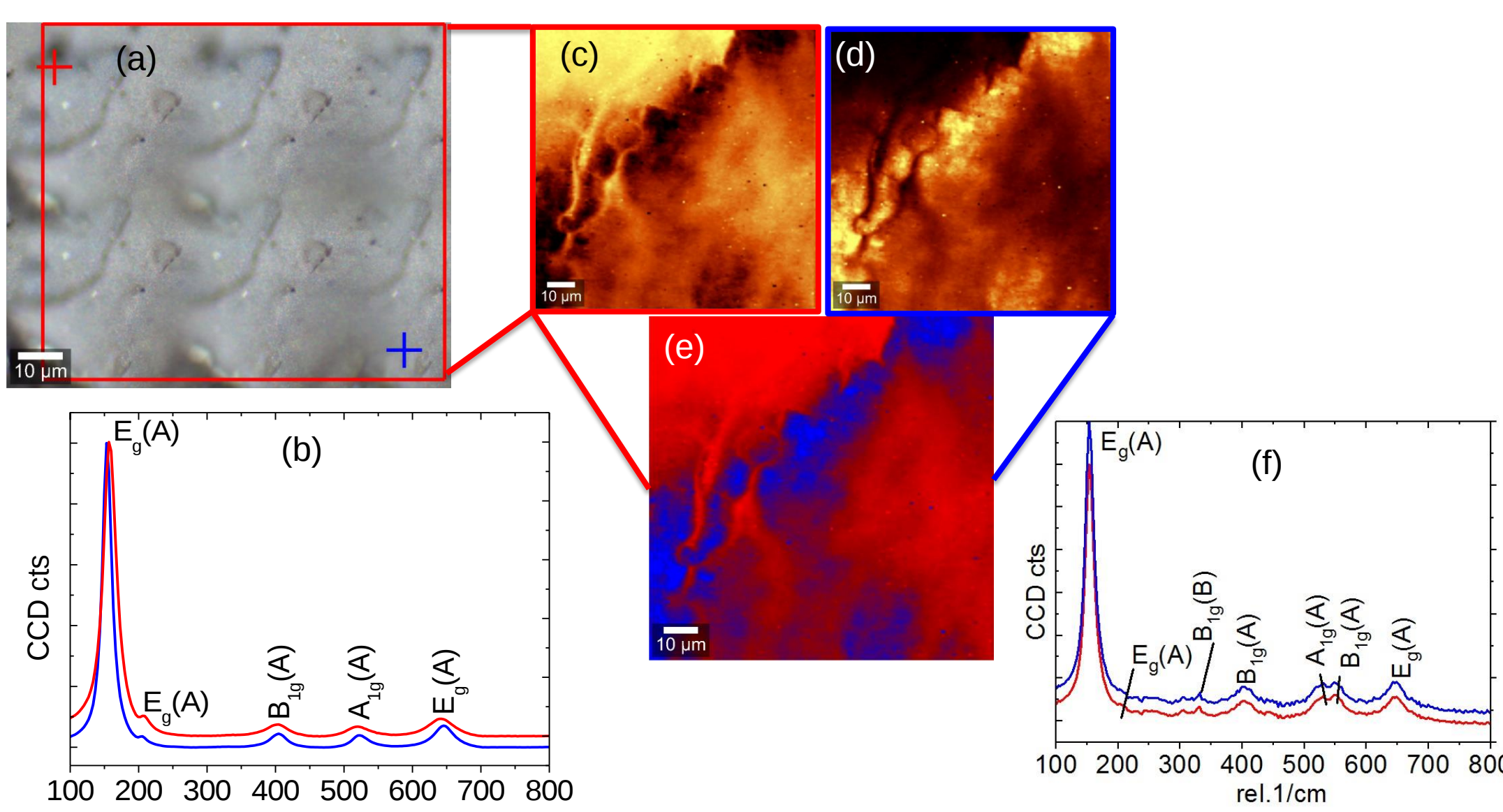
- Raman revealed the distribution of Anatase, Brookite and Rutile phases of TiO₂ in undoped samples.
- Anatase, (Eg₁) at 209.08 cm⁻¹ and 644.59 cm⁻¹(Eg₃).

X-Ray Diffraction



- XRD has revealed peaks at 2θ of 25.49°, 38.11°, 48.14°, 54.58°, 63.00°, 70.33° and 75.66° (Anatase JCPDS No. 21-1272).
- Brookite peak at 40.6° (JCPDS No. 29-1360).
- The lattice constant increases with the increase in carbon concentration along the maximum elongation of c-axis is about 0.687 Å, from 9.143 Å (un-doped TNTs) to 9.830 Å (75 mM TNTs)

Raman Spectroscopy



- The Eg₁ mode of the undoped as TNTs has shown a red shift of Δ9.78 cm⁻¹ relative to the bulk.
- This was attributed to decrease in particle size, presence of Brookite and phonon confinement.
- Upon doping the 9 mM C-TNTs, 27 mM TNTs and 75 mM TNTs have shown a red shift of Δ0.09 cm⁻¹, Δ1.39 cm⁻¹ and 1.81 cm⁻¹ respectively, suggesting incorporation of carbon in the TiO₂ compared to the undoped TiO₂.

Conclusion

- SEM micrographs show changes in morphology of samples as the dopant level increases.
- SEM-EDX has revealed the presence to strong Ti and O signals.
- FTIR has confirmed presence of Ti and O bonds at 580-660 cm⁻¹
- XRD has shown existence of Brookite and Anatase phase TiO₂.
- Raman spectroscopy analysis confirmed the presence of Anatase and Brookite phases of TiO₂.
- The Eg₁ mode of Anatase TiO₂ has shown a red shift as the dopant level increase. Suggesting successful doping of TiO₂, which is in mutual agreement with XRD which has shown increase in lattice parameters along c-axis as the dopant level increases

Acknowledging the support of:

SAIP Conference

July 2017



University of Fort Hare
Together in Excellence